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## **QUANTITATIVE DETERMINATION OF GALLIC ACID AND TANNIC ACID** FROM GALLNUT EXTRACT

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#### **SUMMARY**

Quercus infectoria gallae (oak gall) contain tannins which are characterized to have curative and anti-inflammatory properties. Because of their antiviral and antibacterial qualities, tannins from gallnut extracts have been used in traditional and ayurvedic medicine as well as beauty culture. A newly developed gallnut extract was prepared in a glycerin-water-mixture. To examine the quality of this extract a reliable, innovative HPLC method was worked out to determine the containing active ingredients.

#### INTRODUCTION

Tannins or tanning agents are natural occurring phenolic plant compounds highly abundant in bark, roots, and leaves. Their main operation area is to support the healing process of inflammations, abscesses, incinerations, wounds [4], atopic skin [6] as well as quinsy [1, 5]. The effect of tannins is antibacterial, antiviral [2], antifungal [3] anti-inflammatory, astringent and toxin neutralizing. Tanning agents are divided into three groups: gallotannins, algae tanning agents and catechol tanning agents. Gallic acid, also known as 3,4,5-trihydroxybenzoic acid, is a component of the gallotannins and found highly concentrated in gallnuts and oak bark.

Tannic acid is a specific commercial form of tannin. The chemical formula for tannic acid is often given as C76H52O46, which corresponds with decagalloyl glucose, but in fact it is a mixture of polygalloyl glucoses or polygalloyl quinic acid esters with a varying number of galloyl moieties per molecule.

The following application shows how to determine and quantify gallic acid and tannic acid from gallnut extract with an HPLC method. Since tannic acid was defined as a mixture its determination was carried out as a sum parameter.

#### RESULTS

For the quantitative determination of gallic acid and tannic acid five different measuring points were defined. After calibration the limit of detection (LOD) and the limit of quantification (LOQ) were determined. For gallic acid a LOD of 12 ng/mL and LOQ of 40 ng/mL was achieved. For tannic acid a LOD of 120 ng/mL and LOQ of 400 ng/mL was calculated.

The next step was to measure the sample. The gallnut extract consist of a mixture of glycerol and water and had a strong yellow, almost brown dye. Because of the extract's viscosity a direct injection into the HPLC system was not possible. A dilution series was made and a final dilution with water in a relation of 1:1000 was chosen. The extract was filtered through a 0.45  $\mu$ m pore size hydrophilic filter. For the evaluation of gallic acid and tannic acid pretreated samples with different injection volumes were measured. Gallic acid was analyzed with an injection volume of 10  $\mu$ L whereas an injection volume of 1  $\mu$ L was used for determining the sum parameter tannic acid. **Fig. 1** and **2** show the measurements of the diluted extract at different injection volumes. Furthermore replicates of the filtered and diluted (1:1000) extract were measured with 1  $\mu$ L and 10  $\mu$ L injection volume. The samples are evaluated on the based calibration curves. The replicates show reproducible results. Relating to the detected area the relative standard deviation for the measurements (n=4) is 1.94% RSD for gallic acid and 0.51% RSD for tannic acid.



### **MATERIALS AND METHOD**

An AZURA Analytical HPLC Plus system for a pressure range up to 700 bar was used for this application. It consist of a P 6.1L HPG pump, an autosampler 3950, a CT 2.1 column thermostat and DAD 6.1L. The analytical method runs with a gradient mode at a flow rate of 1 mL/min. The mobile phase is a mixture of water and acetonitrile/water 50:50 (v/v). An amount of 0.1% of formic acid is used as mobile phase modifier. The column thermostat was set to 30 °C and the detector recorded at 280 nm. The used column is filled with KNAUER Eurospher II 100-3 C18H silica.

#### **CONCLUSION**

With this developed method and the AZURA HPLC Plus system it is possible to perform a rapid quantitative analysis of gallic acid and tannic acid without time consuming sample preparation. Despite of the complex matrix like the gallnut extract, the quantification could be performed robustly and reproducibly with the specified method parameters. To exploit the full potency of gallic acid a preparative purification of the extract is possible. For the processing of the purified product it has to be solved in glycerol, water or a mixture of those solvents. Because of the presence of acidic modifier and methanol in the analytical method it cannot be adapted directly up to a preparative dimension. A possible preparative method should be applied immediately after the gallnut extraction and should be run with 100% watery eluent. KNAUER's developed analytical method still can be used for quality and purity control.

#### REFERENCES

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Additional information:

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#### **ADDITIONAL RESULTS**



#### **ADDITIONAL MATERIALS AND METHODS**

Tab. A1 Method parameters

Eluent A

 $H_2O_{dd}$  +0.1 % formic acid



**Fig. A2** Chromatogram Tannic acid,  $\beta$ =0.01 mg/mL

Eluent B	Acetonitrile: H <sub>2</sub> O <sub>dd</sub> 50:50 (v/v) +0.1 % formic acid		
Gradient	Time	% A	% B
	0.00	95	5
	2.00	95	5
	5.00	55	45
	5.02	0	100
	10.00	0	100
	10.02	95	5
	15.00	95	5
Flow rate	1 mL/min	System pressure	_
Column temperature	30 °C	Run time	15 min
Injection volume	1-10 μL	Injection mode	_
Detection wavelength	280 nm	Data rate	20 Hz
		Time constant	0.05 sec

Tab. A2 System configuration & data

Pump	AZURA P 6.1L, HPG, 10 mL, SS	APH35EA
Autosampler	Autosampler 3950	A50070
Detector	AZURA DAD 6.1L	ADC11
Flow cell	LightGuide 50mm, 6µL	AMD59
Thermostat	AZURA CT 2.1	A05852
Eluent tray	AZURA E 2.1L	AZC00
Column	Vertex Plus Column, 150x3mm, Eurospher II 100-3 C18H	15XE185E2G
Software	OpenLAB CDS EZChrom Edition	A2600-1

